# 1-Ethyl-1,4-dihydro-4-oxo-7-(pyridinyl)-3-quinolinecarboxylic Acids. II. Synthesis

Philip M. Carabateas, R. Pauline Brundage, Karl O. Gelotte, Monte D. Gruett, Roman R. Lorenz, Chester J. Opalka, Jr., Baldev Singh, William H. Thielking, Gordon L. Williams and George Y. Lesher\*

Sterling-Winthrop Research Institute, Rensselaer, New York 12144 Received May 23, 1984

The preparation of a variety of 1-ethyl-1,4-dihydro-4-oxo-7-pyridinyl-3-quinolinecarboxylic acids with a range of substituents on the pyridine ring is described. Starting with the appropriately substituted aniline and using the first two steps of the Gould-Jacobs quinoline synthesis the 7-pyridinyl-3-quinolinecarboxylates can be obtained. Ethylation at the 1-position and hydrolysis of the ester group gives the desired acid products. These compounds have significant antibacterial activity: 1-ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylic acid is now in clinical study and 7-(2,6-dimethyl-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinolinecarboxylic acid is under advanced evaluation.

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The finding of exceptional antibacterial activity in 1-ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylic acid **6a** and 7-(2,6-dimethyl-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinolinecarboxylic acid **6g** encouraged us to explore related compounds, especially examining the effect of substituents on the pryidine ring on antibacterial

activity. The preparation of the required intermediate anilinopyridines is described in the preceding paper [1]. Their use to prepare the desired 1-alkyl-1,4-dihydro-4-oxo-7-pyridinyl-3-quinolinecarboxylic acids 6 along with further manipulation of the attached pyridine ring are described in this paper. The biological activity of some of these compounds has been described [2].

#### SCHEME 1

Table 1
Diethyl [[[3-(4-Pyridinyl)phenyl]amino|methylene|propanedioates 3

$$\begin{array}{c} R_6 \\ N \\ R_2 \\ R_2 \\ R_3 \\ \end{array}$$

Compound No.	$R_2$	R <sub>a</sub>	$R_s$	R <sub>6</sub>	Mp (°C)	Yield (%) [a]
3a	Н	Н	Н	Н	89-91	91
3b	CH <sub>3</sub>	H	H	Н	[b]	[c]
<b>3</b> c	$C_2H_5$	H	H	Н	69-71 [ь]	87
<b>3d</b>	H	CH <sub>3</sub>	Н	Н	88-91	76
<b>3</b> e	CH <sub>3</sub>	$CH_3$	Н	Н	74-76	88
3f	CH <sub>3</sub>	Н	CH,	H	86-87	91
<b>3g</b>	CH <sub>3</sub>	Н	Ħ	CH <sub>3</sub>	78-80	78
3h	CH <sub>3</sub>	Н	Н	OCH,	[b]	[c]
<b>3</b> i	$C_2H_5$	H	H	$C_2H_s$	65-67	79
<b>3</b> j	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>	Н	112-114 [b]	86
3k 3ℓ 3m		Py 3-pyridinyl 2-CH₃-5-pyridiny 2-pyridinyl	1		84-86 [b] [b]	95 [e] [c]

[a] All prepared by Procedure A. [b] Not analyzed. [c] Used in the next step without purification.

Table 1a

		Analysis %						
Compound		(	C	1	H	Ī	V	
No.	Formula	Calcd.	Found	Calcd.	Found	Calcd.	Found	
3a	C19H20N2O4	67.05	66.90	5.92	5.97	8.23	8.21	
3d	$C_{20}H_{22}N_2O_4$	67.78	67.72	6.26	6.31	7.91	7.91	
<b>3e</b>	$C_{21}H_{24}N_2O_4$	68.46	68.28	6.57	6.54	7.60	7.80	
3f	$C_{21}H_{24}N_2O_4$	68.46	68.33	6.57	6.52	7.60	7.54	
3g	$C_{21}H_{24}N_2O_4$	68.46	68.50	6.57	6.59	7.60	7.60	
3i	$C_{23}H_{28}N_2O_4$	69.68	69.43	7.12	7.14	7.07	6.91	
3k	$C_{19}H_{20}N_{2}O_{4}$	67.05	67.06	5.92	5.88	8.23	8.23	

Scheme 1 outlines the synthetic path used to convert these anilines 1a-m to the desired quinolonecarboxylic acids 6a-m. The first two steps of the Gould-Jacobs reaction sequence [3] were used to react with the anilines 1 to give the acrylates 3 in good yields (Table 1). These were then cyclized in refluxing Dowtherm A® to the quinolone esters 4 in generally good yields (Tables 2 and 5). Although not looked for carefully, none of the possible isomeric quinolones was detected in these cyclizations. The quinolone esters 4a-m were alkylated with dialkyl sulfates, alkyl iodides or alkyl tosylates in dimethylformamide with potassium carbonate to give 5a-m (Tables 3 and 5). Hydrolysis of 5a-m gave the desired 1-ethyl-1,4-dihydro-4-oxo-7-pyridinyl-3-quinolinecarboxylic acids 6a-m (Tables 4 and 5).

Ethyl has generally been found to be the optimum substituent at the 1-position for maximum antibacterial activity [4]. This was also found to be true in this series but

other lower alkyl groups were also prepared and are listed at the end of Tables 3 and 4. The preparation of the 1-(2-SCHEME 2

Table 2

Ethyl 1,4-Dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylates 4

$$R_2 \xrightarrow{R_6} R_5 \xrightarrow{0} COOC_2H_5$$

Compound No.	$R_z$	$R_3$	$R_s$	R <sub>6</sub>	Mp (°C)	Yield (%) [a]
4a	н	н	н	Н	292-294	75
4b	CH,	н	Н	Н	185-188 [b]	26
4D 4c	C <sub>2</sub> H <sub>5</sub>	H	H	Н	245 dec	59 [c]
4d	H H	CH.	H	Н	[b]	[d]
4e	CH,	CH <sub>3</sub>	Н	H	269-271 [d]	91
4f	CH,	Н	CH.	H	[b]	87
4g	CH.	H	н	CH <sub>3</sub>	> 320 dec	85
4h	CH,	Н	H	OCH <sub>3</sub>	[b]	[d]
4i	$C_2H_5$	Н	H	$C_2H_5$	>300 [b]	90
4j	CH <sub>3</sub>	CH <sub>3</sub>	$CH_3$	H	281-282	45

[a] All prepared by Procedure B. [b] Not analyzed. [c] Hydrochloride salt. [d] Used in the next step without purification.

Table 2a

		Analysis %					
Compound		(	C	1	ł	1	V
No.	Formula	Calcd.	Found	Calcd.	Found	Calcd.	Found
4a	C,7H,4N2O3	69.38	69.17	4.79	4.82	9.52	9.45
4c	$C_{19}H_{18}N_2O_3\cdot HCl$	63.60	63.30	5.34	5.44	7.81	7.78
4g	$C_{19}H_{16}N_{2}O_{3}$	70.79	70.50	5.63	5.59	8.69	8.66
4j	$C_{20}H_{20}N_2O_3$	71.41	71.22	5.99	5.85	8.33	8.10

hydroxyethyl) derivative 5y by direct alkylation of 4a with 2-chloroethanol or ethylene epoxide was unsatisfactory, so we resorted to alkylation with 2-chloroethyl vinyl ether (Scheme 2) giving 5x followed by hydrolytic removal of the vinyl group to give 5y. Hydrolysis of 5y gave the desired 6y. Chlorination of 5y followed by hydrolysis of the ester gave the 1-(2-chloroethyl) acid 6z.

The preparation of the N(py)-oxide of 5a gave the very useful 5n (Scheme 3). It was hydrolysed to the corresponding acid 6n. The N(py)-oxide 5n was rearranged in acetic anhydride to give after hydrolysis the 2-pyridone acid 6o. O-Methylation of 5o with methyl iodide and silver oxide gave 5p which was hydrolysed to the corresponding acid 6p. The chlorination of the N(py)-oxide 5n with phosphorus oxychloride gave the 2-chloropyridine 5q in respectable yield [5]. This was fortunate because attempts to prepare 5q in chlorination of the 2-pyridone 5o were unsuccessful because here chlorination also took place at the 4-position of the quinoline ring with loss of the N-ethyl group, as well as the production of tars. Acid hydrolysis of 5q gave the corresponding acid 6q.

The 2-dimethylaminopyridine derivatives **5r** and **6r** were also obtained from **5q**. Teatment of **5n** with dimethyl

# **SCHEME 3**

Table 3

Ethyl 1-Alkyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylates 5

$$\begin{array}{c|c} R_8 & R_5 & O \\ \hline N & R_2 & N \\ R_2 & R_3 & D \\ \hline \end{array}$$

Compou	nd				K <sub>1</sub>			
No.	$R_z$	$R_3$	$R_5$	$R_6$		Mp (°C)	Yield (%)	Procedure
5a	$C_2H_5$	Н	Н	Н		169-170.5	45	С
5b	$C_2H_5$	CH <sub>3</sub>	Н	Н	Н	[a]	[b]	С
5c	$C_2H_5$	$C_2H_5$	H	H	Н	143-144 [a]	73	С
5d	$C_2H_5$	H	$CH_3$	Н	Н	162-163	16	С
5e	$C_2H_5$	$CH_3$	$CH_3$	Н	Н	101-103	23	С
5f	$C_2H_5$	$CH_3$	H	$CH_3$	Н	157-158 [a]	46	С
5g	$C_2H_5$	CH <sub>3</sub>	H	Н	CH <sub>3</sub>	168-170	65	С
5h	$C_2H_5$	CH <sub>3</sub>	H	Н	OCH <sub>3</sub>	171-173	29	Ċ
5i	$C_2H_5$	$C_2H_5$	H	Н	$C_2H_5$	135-137 [a]	77	C
5j	$C_2H_5$	CH <sub>3</sub>	$CH_3$	CH <sub>3</sub>	H	226-228 [a]	91	C
5o	$C_2H_5$	ОН	Н	H	H	263-266 [a]	53	_
5р	$C_2H_5$	$OCH_3$	H	Н	H	196-197	65	_
5q	$C_2H_5$	Cl	H	H	Н .	164-166 [a]	70	_
5 <b>r</b>	$C_2H_5$	$N(CH_3)_2$	H	Н	Н	[a]	[b]	
5s	$C_2H_5$	CN	Н	Н	H	258-261	62	
5v	CH <sub>3</sub>	Н	Н	Н	H	[a]	[b]	С
5w	$C_3H_7$	Н	Н	Н	Н	[a]	[b]	C
5x	$CH_2CH_2OCH = CH_2$	Н	Н	Н	Н	191-193 [a]	61	C
5 <b>y</b>	CH <sub>2</sub> CH <sub>2</sub> OH	Н	Н	Н	H	154-156 [a]	74	_
5z	CH <sub>2</sub> CH <sub>2</sub> Cl	Н	Н	Н	Н	205 dec [a]	72	_
5aa	CH <sub>3</sub>	CH <sub>3</sub>	Н	Н	CH <sub>3</sub>	215-217 [a]	82	С
5ab	$C_3H_7$	$CH_3$	Н	Н	CH <sub>3</sub>	135-136 [a]	76	С

[a] Not analyzed. [b] Used in the next step without purification.

Table 3a

				Analy	sis %		
Compound	d	(	3	I	H	]	N
No.	Formula	Calcd.	Found	Calcd.	Found	Calcd.	Found
5a	$C_{19}H_{18}N_2O_3$	70.79	70.76	5.63	5.71	8.69	8.46
5d	$C_{20}H_{20}N_2O_3$	71.41	71.19	5.99	5.92	8.33	8.68
5e	$C_{21}H_{22}N_{2}O_{3}$	71.98	71.69	6.33	6.29	7.99	7.89
5g	$C_{21}H_{22}N_2O_3$	71.98	71.99	6.33	6.37	7.99	7.99
5h	$C_{21}H_{22}N_{2}O_{4}$	68.84	68.46	6.05	6.02	7.65	7.66
5p	$C_{20}H_{20}N_{2}O_{4}$	68.17	68.01	5.72	5.66	7.95	7.97
5s	$C_{20}H_{17}N_3O_3$	69.15	69.04	4.93	4.96	12.10	12.29

sulfate and then potassium cyanide (Scheme 3) gave the 2-cyanopyridine **5s** [6]. This was converted to the diacid **6s** by basic hydrolysis.

Scheme 3 also outlines the conversion of the 2,6-dimethylpyridine  $\mathbf{5g}$  to its N(py)-oxide  $\mathbf{5t}$  which was hydrolysed to the acid  $\mathbf{6t}$ , as well as rearranged in acetic anhydride followed by hydrolysis, to the 2-hydroxymethyl-6-methylpyridine derivative  $\mathbf{6u}$ .

After further biological evaluation **6a** was chosen for clinical studies as an agent for gonorrheal infections and urinary tract infections, as well as for systemic gram-negative

infections [2]. Its generic name is rosoxacin [7].

The 2,6-dimethylpyridinyl derivative **6g** was found to have surprisingly potent antistaphylococcal activity and is being developed for this indication.

# **EXPERIMENTAL**

Melting points are uncorrected. All new compounds have compatible ir, ms and/or nmr spectra.

Procedure A. Diethyl [[[3-(4-Pyridinyl)phenyl]amino]methylene]propanedioate (3a).

A mixture of 4-(3-aminophenyl)pyridine (1a) (112 g, 0.66 mole) and diethyl ethoxymethylenemalonate (2) (148 g, 0.68 mole) was heated at 130° for eight minutes and then allowed to cool to 100°. The resulting oil was taken up in 100 ml of 2-propanol and cyclohexane was added to incipient cloudiness. This was treated with charcoal, filtered and cooled to yield 205 g (91%) of adduct 3a, mp 86-88°. An analytical sample was obtained by recrystallization from ethanol, mp 89-91°.

Table 4

1-Alkyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylic Acids 6

Compound No.	$R_2$	$R_3$	$R_s$	$R_6$		Mp (°C)	Yield (%) [a]
6a	$C_2H_5$	Н	Н	Н	Н	284-286	89
6b	$C_2H_5$	CH,	Н	H	Н	302-303 dec	76 [b]
6c	$C_2H_5$	$C_2H_5$	Н	H	Н	> 300	87 [b]
6d	$C_2H_5$	Н	CH <sub>3</sub>	Н	Н	284-287 dec	62 [b]
6e	$C_2H_5$	CH <sub>3</sub>	CH <sub>3</sub>	H	Н	326-328 dec	64
6f	$C_2H_5$	СН,	Н	CH <sub>3</sub>	Н	> 300	87 [b]
6g	$C_2H_5$	CH <sub>3</sub>	Н	Н	CH <sub>3</sub>	$282-284  \deg$	24
6h	$C_2H_5$	CH <sub>3</sub>	Н	Н	OCH <sub>3</sub>	277-278	81
6i	$C_2H_5$	$C_2H_5$	Н	Н	$C_2H_5$	210-212	89
6j	$C_2H_5$	CH <sub>3</sub>	CH <sub>3</sub>	CH <sub>3</sub>	H	277-279 dec	72
<b>6</b> 0	$C_2H_5$	ОН	Н	Н	Н	352-354	69
6р	$C_2H_5$	OCH <sub>3</sub>	Н	Н	H	238-240	75
6q	$C_2H_5$	Cl	Н	H .	Н	264-265 dec	66
6r	$C_2H_5$	$N(CH_3)_2$	Н	Н	Н	225-245 dec	19 [b]
6s	$C_2H_5$	COOH	Н	H	H	246-247	59
6u	$C_2H_5$	CH <sub>3</sub>	Н	H	CH <sub>2</sub> OH	250-251	64
6v	CH <sub>3</sub>	Н	H	H	Н	329-330 dec	38
6w	$C_3H_7$	Н	Н	Н	Н	295-297 dec	42
6y	CH <sub>2</sub> CH <sub>2</sub> OH	Н	Н	H	Н	285-287	68
6z	CH <sub>2</sub> CH <sub>2</sub> Cl	Н	Н	Н	Н	> 300	69 [b]
6aa	CH <sub>3</sub>	CH <sub>3</sub>	Н	Н	Н	> 335	85 [b]
6ab	$C_3H$ ,	CH <sub>3</sub>	Н	Н	CH <sub>3</sub>	233-235	77

[a] Prepared by Procedure D. [b] Hydrochloride salt.

Table 4a

		Analysis %					
Compound	1		3	_	I	1	
No.	Formula	Calcd.	Found	Calcd.	Found	Calcd.	Found
6a	$C_{17}H_{14}N_{2}O_{3}$	69.38	69.33	4.79	4.80	9.52	9.59
6b	C <sub>18</sub> H <sub>16</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	62.70	62.50	4.97	4.97	8.12	8.09
6c	C <sub>19</sub> H <sub>18</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	63.60	63.48	5.34	5.33	7.81	7.71
6d	C <sub>18</sub> H <sub>16</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	62.70	63.03	4.97	4.91	8.12	8.24
6e	$C_{19}H_{18}N_2O_3$	70.79	70.63	5.63	5.59	8.69	8.71
6f	$C_{19}H_{18}N_2O_3\cdot HCl$	63.60	63.55	5.34	5.23	7.81	7.66
6g	$C_{19}H_{18}N_2O_3$	70.79	70.70	5.63	5.53	8.69	8.88
6h	$C_{19}H_{18}N_{2}O_{4}$	67.45	67.23	5.36	5.35	8.28	8.61
6i	$C_{21}H_{22}N_{2}O_{3}$	71.98	71.80	6.33	6.24	7.99	7.86
6j	$C_{20}H_{20}N_{2}O_{3}$	71.41	71.21	5.99	5.93	8.33	8.16
<b>6</b> 0	$C_{17}H_{14}N_{2}O_{4}$	65.80	65.76	4.55	4.50	9.03	9.12
6р	$C_{18}H_{16}N_{2}O_{4}$	66.66	66.72	4.97	5.31	8.64	8.64
6q	$C_{17}H_{13}CIN_2O_3$	62.11	61.97	3.99	3.99	8.52	8.91
6r	$C_{19}H_{19}N_3O_3\cdot HCl$	61.04	60.82	5.39	5.29	11.24	11.06
<b>6s</b>	$C_{18}H_{14}N_{2}O_{5}$	63.90	63.70	4.17	4.26	8.20	8.21
6u	$C_{19}H_{18}N_{2}O_{4}$	67.45	67.66	5.36	5.31	8.28	8.28
6v	$C_{16}H_{12}N_2O_3$	68.57	68.44	4.32	4.35	9.99	9.96
6w	$C_{18}H_{16}N_{2}O_{3}$	70.12	70.06	5.23	5.22	9.09	9.09
6y	$C_{17}H_{14}N_{2}O_{4}$	65.80	65.75	4.55	4.54	9.03	9.22
6z	C <sub>17</sub> H <sub>13</sub> ClN <sub>2</sub> O <sub>3</sub> ·HCl	55.90	55.96	3.86	3.86	7.67	7.71
6aa	$C_{18}H_{16}N_2O_3$ ·HCl	62.07	62.47	4.97	4.86	8.12	8.09
6ab	$C_{20}H_{20}N_2O_3$	71.41	71.05	5.99	5.92	8.33	8.27

Procedure B. Ethyl 1,4-Dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylate (4a).

Dowtherm A® (500 ml) was heated to boiling with stirring. Diethyl [[[3-(4-pyridinyl)phenyl]amino]methylene]propanedioate (3a) (50 g, 0.15 mole) was added all at once and the solution was boiled vigorously with stirring for 12 minutes. After the solution had cooled to room temperature, it was diluted with one volume of n-hexane and the resulting tan solid was collected and washed well with ether. The product (32.7 g, 75%) melted at 242-245°. This material was satisfactory for further use. An analytical sample was obtained by recrystallization from dimethylformamide, collecting the crystals at 60-70° to avoid gel formation which occurs on further cooling. The product 4a was washed with ethanol and dried, mp 292-294°.

Procedure C. Ethyl 1-Ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinoline-carboxylate (5a).

A mixture of dimethylformamide (1  $\ell$ ), potassium carbonate (milled, 165 g, 1.2 mole) and ethyl 1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinoline-carboxylate (4a) (100 g, 0.34 mole) was stirred at 60° for 15 minutes. Diethyl sulfate (52.5 g, 0.34 mole) was added and the mixture stirred for one hour. The dimethylformamide was evaporated and the resulting residue was partitioned between water and chloroform (emulsion, broken by filtration). The chloroform was evaporated and the dark residual oil was taken up in 2-propanol (1  $\ell$ ), treated with charcoal and cooled. An orange solid (49.0 g, 45%) was collected. An analytical sample of 5a was obtained by recrystallization from 2-propanol, mp 169-170.5°.

Table 5
1,4-Dihydro-4-oxo-7-pyridinyl-3-quinolinecarboxylates

Compound						
No.	Py	$R_1$	R	Mp (°C)	Yield (%)	Procedure
4k	3-pyridinyl	77	CH	050 056 5 1	05	
		Н	$C_2H_s$	253-256 [a]	85	В
5k	3-pyridinyl	$C_2H_5$	$C_2H_5$	[a]	[b]	С
6k	3-pyridinyl	$C_2H_5$	H	268-269	39	D
<b>4</b> ℓ	2-CH <sub>3</sub> -5-pyridinyl	Н	$C_2H_5$	[6]	n.i	В
<b>5</b> ℓ				[a]	[b]	
	2-CH <sub>3</sub> -5-pyridinyl	$C_2H_5$	$C_2H_5$	[a]	[b]	С
<b>6</b> ℓ	2-CH <sub>3</sub> -5-pyridinyl	$C_2H_5$	Н	277-280 dec	26 [c] [d]	D
4m	2-pyridinyl	н	$C_2H_5$	[a]	[b]	В
5m	2-pyridinyl	$C_zH_s$	$C_2H_5$	[a]	[b]	Č
6m	2-pyridinyl	$C_2H_5$	H	219-220	35	Ď
VIII	2-pyridinyi	C <sub>2</sub> 11 <sub>5</sub>	11 -	219-220	33	D
5n	4-pyridinyl N-oxide	$C_2H_5$	$C_2H_5$	222-224	61	С
6n	4-pyridinyl N-oxide	C <sub>2</sub> H <sub>5</sub>	H	307-309 dec	80	D
5t	2.6 (CU ) 4 monidical Nid-	C II	СП	020 041	60	
	2,6-(CH <sub>3</sub> ) <sub>2</sub> -4-pyridinyl N-oxide	C <sub>2</sub> H <sub>5</sub>	$C_2H_5$	238-241	68	С
6t	$2,6-(CH_3)_2-4$ -pyridinyl N-oxide	$C_2H_5$	Н	295-296 dec	64	D

[a] Not analyzed. [b] Used in next step without purification. [c] Hydrochloride salt. [d] Yield is for alkylation and hydrolysis steps.

Table 5a

				Analy	sis %		
Compou	nd	(	3	ŀ	H	I	V
No.	Formula	Calcd.	Found	Calcd.	Found	Calcd.	Found
6k	C,,H,,N,O,	69.38	69.07	4.79	4.74	9.52	9.72
<b>6</b> ℓ	C <sub>18</sub> H <sub>16</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	62.70	62.42	4.97	5.07	8.12	8.11
6m	$C_{17}H_{14}N_{1}O_{3}$	69.38	69.53	4.79	4.78	9.52	9.35
5n	$C_{10}H_{10}N_{2}O_{4}$	67.45	67.29	5.36	5.35	8.28	8.18
6n	$C_{17}H_{14}N_{2}O_{4}$	65.80	65.72	4.55	4.48	9.03	9.00
5t	C, H, N, O,	68.84	68.73	6.05	6.09	7.65	7.84
6t	$C_{19}H_{18}N_2O_4$	67.45	67.19	5.36	5.41	8.28	8.33

Procedure D. 1-Ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylic Acid (6a).

A mixture of ethyl 1-ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinoline-carboxylate (5a) (41.0 g, 0.13 mole) and 400 ml of 10% potassium hydroxide solution was stirred and heated on a steam bath for one hour. The resulting solution was treated with charcoal, filtered and the filtrate acidified to pH 6 with acetic acid. The suspension was cooled and the product 6a (33.0 g, 89%) was collected. An analytical sample was obtained by crystallization from dimethylformamide, mp 284-286°.

Ethyl 1-[2-(Ethenyloxy)ethyl]-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinoline-carboxylate (5x).

Ethyl 1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylate (4a) (37 g, 0.13 mole) was stirred together with potassium carbonate (milled, 28 g, 0.2 mole) in dimethylformamide (250 ml) on a steam bath for one hour. 2-Chloroethyl vinyl ether (13.8 g, 0.13 mole) was then added and heating was continued for 3.5 hours. The hot mixture was filtered, the filtrate

chilled, and the resulting solid was collected, washed with ethanol and dried (10.5 g). The filtrate was heated on a steam bath for an additional 15 hours and again chilled to obtain a second crop of product (5.5 g). The two crops were combined and recrystallized from dimethylformamide to give 26.2 g (61%) of product 5x, mp 191-193°.

Ethyl 1,4-Dihydro-1-(2-hydroxyethyl)-4-oxo-7-(4-pyridinyl)-3-quinoline-carboxylate (5y).

The vinyl ether 5x (26.2 g, 0.07 mole) was dissolved in acetic acid (50 ml) with warming. Water (5 ml) was added and the solution heated on a steam bath for one hour. The solution was concentrated and the residual solid was triturated with water, washed with ethanol, dried and recrystallized from dimethylformamide to give 18.0 g (74%) of product 5y, mp 154-156°.

1,4-Dihydro-1-(2-hydroxyethyl)-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylic Acid (6y).

The ester 5y (4.2 g, 0.012 mole) was heated on a steam bath with 5% potassium hydroxide (30 ml) for two hours with stirring. The solution was neutralized ( $pH \sim 6$ ) with dilute hydrochloric acid. The resulting precipitate was collected, recrystallized twice from dimethylformamide and dried giving 2.6 g (68%) of product 5y, mp 285-287°.

1-(2-Chloroethyl)-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylic Acid Monohydrochloride (6z).

The ester 5y (15.0 g, 0.044 mole) was suspended in 800 ml of dry chloroform and thionyl chloride (17 ml) was added with stirring, a tar formed. The mixture was heated on a steam bath with stirring for five hours and the tar solidified. This solid was collected, dried and suspended in water (250 ml). The mixture was slowly neutralized ( $pH \sim 7.5$ ) with 10% potassium hydroxide, and resulting solid was collected and recrystallized twice from ethanol to give 11.4 g (72%) of a tan solid of im-

precise melting point (ca 205°). Its nmr spectrum confirmed that this was reasonably pure ethyl 1-(2-chloroethyl)-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylate. This ester was heated with stirring on a steam bath with 300 ml of 2N hydrochloric acid for 1.5 hours. The mixture was filtered and the filter cake was triturated with water, washed with ethanol and dried to give 11.1 g (96%; yield from both steps, 69%) of the product 6z, mp > 300°.

Ethyl 1-Ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylate N(py)-Oxide (5n).

To a stirred solution of ethyl 1-ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinolinecarboxylate (5a) (18.0 g, 0.056 mole) in 250 ml of methylene dichloride was added 85% m-chloroperoxybenzoic acid (12.2 g, 0.06 mole). A white solid crystallized after one hour. Stirring was continued for one more hour. The m-chlorobenzoic acid salt of the N-oxide product was collected and then boiled five minutes in a mixture of 25 ml of dimethylformamide and 100 ml of ethanol. This suspension was cooled, the solid collected and stirred for ten minutes with a warm sodium bicarbonate solution. The white solid was collected, washed with water, dried and recrystallized from ethanol. The product 5n (11.5 g, 61%) melted at 222-224°.

Ethyl 1-Ethyl-1,4-dihydro-7-(1,2-dihydro-2-oxo-4-pyridinyl)-4-oxo-3-quinolinecarboxylate (50).

A mixture of the N(py)-oxide 5n (46.0 g, 0.136 mole) and acetic anhydride (460 ml) was refluxed for eight hours. The green solution was evaporated to a green gum. The gum was stirred with ethanol and the gray-green solid collected. The solid was recrystallized from ethanol and then from dimethylformamide. This was then washed with acetonitrile and recrystallized again from ethanol. The product 5o (24.3 g, 53%) melted at 263-266°.

Ethyl 1-Ethyl-1,4-dihydro-7-(2-methoxy-4-pyridinyl)-4-oxo-3-quinoline-carboxylate (5p).

A mixture of ethyl 1-ethyl-1,4-dihydro-7-(1,2-dihydro-2-oxo-4-pyridinyl)-4-oxo-3-quinolinecarboxylate (50) (19.0 g, 0.056 mole), silver carbonate (31.0 g, 0.112 mole) and 500 ml tetrahydrofuran was heated to reflux with stirring. Methyl iodide (16.1 g, 0.113 mole) was added during five minutes. After three hours of reflux, another 10 ml of methyl iodide was added and refluxing was continued 15 hours longer. After the mixture was cooled the solid was collected and the filtrate evaporated to a yellow solid. This solid was taken up in chloroform, this was filtered through a short column of Florisil and the filtrate evaporated. The resulting solid was recrystallized twice from acetonitrile and then from dioxane to give 12.8 g (65%) of product 5p, mp 196-197°.

Ethyl 7-(2-chloro-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinolinecarboxylate (5q).

A mixture of ethyl 1-ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinoline-carboxylate N(py)-oxide (5n) (15.0 g, 0.044 mole) and phosphorus oxychloride (200 ml) was heated eight hours on the steam bath. The excess phosphorus oxychloride was evaporated and the residue partitioned between saturated sodium bicarbonate solution and chloroform. The organic phase was separated and evaporated to give 13.5 g of a red gum which was crystallized from 100 ml of ethanol to give 11.0 g (70%) of a pale pink solid 5q, mp 164-166°.

7-[2-(Dimethylamino)-4-pyridinyl]-1-ethyl-1,4-dihydro-4-oxo-3-quinoline-carboxylic Acid Monohydrochloride (6r).

A solution of ethyl 7-(2-chloro-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinolinecarboxylate (5q) (5.0 g, 0.014 mole) in 100 ml of ethanol and 50 ml of dimethylamine was heated to 150° for eight hours in a rocking autoclave. The solution was evaporated to a dark oil which solidified. This solid was combined with 50 ml of methanol, 35 ml of 35% sodium hydroxide and 100 ml of water and then heated on a steam bath for two hours. The solution was evaporated and the residue was taken up in a little water and neutralized with dilute hydrochloric acid. The solid was collected, dissolved in methanol and treated with 5 ml of concentrated hydrochloric acid. The cream solid which formed was collected and re-

crystallized from boiling 6N hydrochloric acid. The product 6r (1.0 g, 19%) melted at 225-245° dec.

Ethyl 7-(2-Cyano-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinolinecarboxylate (56).

A mixture of ethyl 1-ethyl-1,4-dihydro-4-oxo-7-(4-pyridinyl)-3-quinoline-carboxylate N(py)-oxide (5n) (27.0 g, 0.08 mole), dimethyl sulfate (20.2 g, 0.16 mole) and 75 ml of acetonitrile was refluxed with stirring for 30 minutes, giving a clear solution. The solution was cooled to 25° and 30 g of potassium cyanide in 100 ml of water was added rapidly with vigorous stirring. A thick white paste formed. After ten minutes the mixture was diluted with one volume of water, stirred a few minutes and the solid collected and washed with water. The dried solid was recrystallized twice by dissolving in hot dimethylformamide, filtering, adding acetonitrile till cloudy and cooling. The product 5s (17.2 g, 62%) melted at 258-261°.

7-(2-Carboxy-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinolinecarboxylic Acid (6s).

A mixture of ethyl 7-(2-cyano-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinolinecarboxylate (5s) (7.0 g, 0.02 mole), 30 ml of water and 20 ml of 10% sodium hydroxide solution was refluxed for three hours. The solution was acidified with dilute hydrochloric acid, the resulting white solid was collected and washed with water. The dried solid was recrystallized from dimethylformamide-ethanol. The resulting solid was dissolved in dilute ammonium hydroxide, the solution was filtered and then acidified with dilute hydrochloride acid. The white solid was washed with water and dried. The product 6s (4.0 g, 59%) melted at 246-247°.

Ethyl 7-(2,6-Dimethyl-4-pyridinyl)-1-ethyl-1,4-dihydro-4-oxo-3-quinoline-carboxylate N(py)-Oxide (5t).

This compound was prepared from 5g in the same manner as 5n in 68% yield, mp 238-241°.

1-Ethyl-1,4-dihydro-7-[2-(hydroxymethyl)-6-methyl-4-pyridinyl]-4-oxo-3quinolinecarboxylic Acid (6u).

Acetic anhydride (30 ml) was heated to boiling, the heat was removed and the N(py)-oxide 5t (10.0 g, 0.027 mole) was added in portions with stirring during five minutes. The dark solution was refluxed for one-half hour, concentrated to a gum which was then refluxed one hour in 50 ml of 6N hydrochloric acid. The solution was concentrated, resulting in a gum which when triturated with ethanol gave a red solid. The solid was taken up in water, treated with charcoal and the resulting filtrate treated with a saturated sodium acetate solution. The yellow solid that formed was collected, air dried and recrystallized successively from acetonitrile, methanol and acetonitrile. The product 6u (5.9 g, 64%) melted at 250-251°.

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